

Optical absorption spectra, crystal-field energy levels and intensities of Eu^{3+} in $\text{GdAl}_3(\text{BO}_3)_4$

This article has been downloaded from IOPscience. Please scroll down to see the full text article.

1994 J. Phys.: Condens. Matter 6 7797

(<http://iopscience.iop.org/0953-8984/6/38/017>)

View [the table of contents for this issue](#), or go to the [journal homepage](#) for more

Download details:

IP Address: 171.66.16.151

The article was downloaded on 12/05/2010 at 20:35

Please note that [terms and conditions apply](#).

Optical absorption spectra, crystal-field energy levels and intensities of Eu^{3+} in $\text{GdAl}_3(\text{BO}_3)_4$

C Görller-Walrand, E Huygen, K Binnemans and L Fluyt

University of Leuven, Laboratory of Inorganic Chemistry, Celestijnenlaan 200F, B-3001 Heverlee, Belgium

Received 4 March 1994, in final form 6 May 1994

Abstract. Locations and assignments of crystal-field levels in polarized absorption spectra (77 and 298 K) are reported for Eu^{3+} in the $\text{GdAl}_3(\text{BO}_3)_4$ huntite matrix. The levels are analysed in terms of 20 free-ion and six crystal-field parameters (D_3 symmetry). J mixing is taken into account. The influence of a parameter change on the calculated energy levels is discussed and it is shown that some parameters affect only particular levels. Spectral intensities of the 4f–4f transitions are given and parametrized using a set of $B_{\lambda k q}$ intensity parameters.

1. Introduction

The crystal structure of $\text{GdAl}_3(\text{BO}_3)_4$ doped with Eu^{3+} is reported by Kuroda *et al* [1]. The host lattice is isostructural with the mineral huntite, $\text{CaMg}_3(\text{CO}_3)_4$. There is only one crystallographic site available for the lanthanide ions. Eu^{3+} is surrounded by six oxygens and the coordination polyhedron is a distorted trigonal prism. Because the upper triangle is twisted over an angle ϕ with regard to the lower triangle, the site symmetry is lowered by this distortion from D_{3h} to D_3 . The first spectroscopic data of Eu^{3+} in $\text{GdAl}_3(\text{BO}_3)_4$ (polarized emission spectra) have been published by Lagerwey and Blasse [2]. They found that the $^5D_0 \rightarrow ^7F_1$ transition is strongly polarized, whereas the $^5D_2 \rightarrow ^7F_0$ transition is unpolarized. The authors consider the latter transition as a pseudoquadrupole transition. Their work has been repeated by Peacock [3], but this author finds that the $^5D_2 \rightarrow ^7F_0$ transition is polarized according to the selection rules for induced electric dipole transitions in a D_3 symmetry. He also published the axial and polarized absorption spectra of the $^5D_1 \leftarrow ^7F_0$ and $^5D_2 \leftarrow ^7F_0$ transitions. The intensities of the $^5D_1 \leftarrow ^7F_0$, $^5D_2 \leftarrow ^7F_0$, $^5D_0 \leftarrow ^7F_1$ and $^5D_1 \leftarrow ^7F_1$ transitions are discussed by Kuroda *et al* [1] in the framework of anisotropic ligand polarization contributions to the transition probabilities in Eu^{3+} . Kellendonk and Blasse [4] have reported the emission and excitation spectra of the Eu^{3+} luminescence and energy transfer phenomena in $\text{Gd}_{1-x}\text{Eu}_x\text{Al}_3(\text{BO}_3)_4$ ($0 < x \leq 1$). The position of the energy levels between 0 and 19 000 cm^{-1} of Eu^{3+} in $\text{EuAl}_3(\text{BO}_3)_4$ are given. The axial absorption, polarized absorption, fluorescence and magnetic circular dichroism (MCD) spectra of Eu^{3+} -doped $\text{YAl}_3(\text{BO}_3)_4$ have been published by Görller-Walrand *et al* [5–7].

In this paper we report the polarized absorption spectra of Eu^{3+} -doped $\text{GdAl}_3(\text{BO}_3)_4$ between 16 000 and 32 000 cm^{-1} at 77 and 298 K. The data are analysed using a parametric Hamiltonian with a total of 26 parameters. The influence of a parameter change on the calculated energy levels is considered. It will be shown that a change does not have a similar effect on each level. The dipole strengths of the 4f–4f transitions in Eu^{3+} are given

and parametrized, on the basis of the anisotropic ligand polarization model proposed by Mason *et al* [8], and adapted by Reid and Richardson [9–11].

2. Experimental details

The Eu^{3+} -doped $\text{GdAl}_3(\text{BO}_3)_4$ crystals were kindly donated by Professor G Blasse of Utrecht University (The Netherlands). Optical absorption spectra were recorded using an AVIV 17DS spectrophotometer. In the visible and UV regions, the instrument has a wavelength resolution better than 0.1 nm. The wavelength accuracy is ± 0.4 nm. Light polarization is achieved by a Glan–Thompson polarizer. For low-temperature measurements, the sample is cooled in an optical dewar (Oxford Instruments) filled with liquid nitrogen (77 K).

3. Selection rules

The crystal-field levels are assigned in a D_3 symmetry and they have the symmetry labels Γ_1 , Γ_2 and Γ_3 , according to the Koster notations [12]. The assignments are based on the polarization characteristics of the transitions and the selection rules for induced electric dipole (ED) and magnetic dipole (MD) transitions (table 1); α , σ and π are defined in the usual manner as

$$\alpha \text{ spectrum: } c \parallel z$$

$$\sigma \text{ spectrum: } c \perp z, \mathbf{H} \parallel c$$

$$\pi \text{ spectrum: } c \perp z, \mathbf{E} \parallel c$$

z is the propagation direction of the light, c is the main crystal axis, and \mathbf{E} and \mathbf{H} are the electric and magnetic field vectors of the incident light respectively.

Table 1. Selection rules for induced electric dipole (ED) and magnetic dipole (MD) transitions in a D_3 symmetry.

	ED			MD		
	Γ_1	Γ_2	Γ_3	Γ_1	Γ_2	Γ_3
Γ_1	—	π	α, σ	—	σ	α, π
Γ_2	π	—	α, σ	σ	—	α, π
Γ_3	α, σ	α, σ	α, σ, π	α, π	α, π	α, π, σ

The ground state of Eu^{3+} is 7F_0 and it has symmetry label Γ_1 . Because no $\Gamma_1 \leftarrow \Gamma_1$ transitions are allowed by the selection rules, transitions starting from 7F_0 are not suitable for detecting crystal-field levels with Γ_1 symmetry. These levels can be located with the aid of transitions starting from the 7F_1 multiplet (which has crystal-field levels with Γ_2 and Γ_3 symmetry).

4. Analysis of the polarized spectra

4.1. ${}^5D_0 \leftarrow {}^7F_{0,1,2}$

Transitions to 5D_0 are found between 16 000 and 17 000 cm^{-1} .

The ${}^5D_0 \leftarrow {}^7F_0$ transition is symmetry-forbidden for both ED and MD transitions ($\Gamma_1 \leftarrow \Gamma_1$) and is not observed in the spectra.

For ${}^5D_0 \leftarrow {}^7F_1$, two absorption peaks are observed, one in σ polarization at 16 787 cm^{-1} ($\Gamma_1 \leftarrow \Gamma_2$) and another in both α and π polarization at 16 915 cm^{-1} ($\Gamma_1 \leftarrow \Gamma_3$). According to the selection rules, it is a MD transition. The energy difference between the Γ_3 and the Γ_2 levels of 7F_1 is 128 cm^{-1} .

The ${}^5D_0 \leftarrow {}^7F_2$ transition shows two absorption peaks in the α and σ spectrum: 16 184 cm^{-1} ($\Gamma_1 \leftarrow \Gamma_3^b$) and 16 305 cm^{-1} ($\Gamma_1 \leftarrow \Gamma_3^a$). No peaks are observed in the π spectrum. This transition has an ED character. The energy gap between the two Γ_3 levels of the 7F_2 multiplet is 121 cm^{-1} .

4.2. ${}^5D_1 \leftarrow {}^7F_{0,1,2}$

These transitions can be found in the spectrum between 17 900 and 19 100 cm^{-1} .

The ${}^5D_1 \leftarrow {}^7F_0$ transition shows two peaks in the spectrum, one in both α and π polarization at 19 000 cm^{-1} ($\Gamma_3 \leftarrow \Gamma_1$) and one in σ polarization at 19 030 cm^{-1} ($\Gamma_2 \leftarrow \Gamma_1$). This transition is analogous to the ${}^5D_0 \leftarrow {}^7F_1$ transition and also has a MD nature. The energy difference between the two crystal-field levels of the 5D_1 multiplet (Γ_2 and Γ_3) is 30 cm^{-1} .

For the ${}^5D_1 \leftarrow {}^7F_1$ three peaks are detected in the α and σ spectrum: 18 539 cm^{-1} ($\Gamma_3 \leftarrow \Gamma_2$), 18 667 cm^{-1} ($\Gamma_3 \leftarrow \Gamma_3$) and 18 700 cm^{-1} ($\Gamma_2 \leftarrow \Gamma_3$). The π spectrum shows no signals. Although $\Delta J = 0$, ${}^5D_1 \leftarrow {}^7F_1$ is a MD transition according to the selection rules. With the knowledge of the position of the 5D_1 crystal-field levels from ${}^5D_1 \leftarrow {}^7F_0$, the crystal-field levels of 7F_1 can be determined from the ${}^5D_1 \leftarrow {}^7F_1$ transition: Γ_3 at 331 cm^{-1} and Γ_2 at 461 cm^{-1} . The 5D_0 level can be placed at 17 245 cm^{-1} .

The ${}^5D_1 \leftarrow {}^7F_2$ signals are very weak, and only peaks in σ polarization can be assigned without ambiguity: 17 938 cm^{-1} ($\Gamma_3 \leftarrow \Gamma_3^b$), 17 969 cm^{-1} ($\Gamma_2 \leftarrow \Gamma_3^b$), 18 058 cm^{-1} ($\Gamma_3 \leftarrow \Gamma_3^a$) and 18 090 cm^{-1} ($\Gamma_2 \leftarrow \Gamma_3^a$). These transitions make it possible to find the energy of some crystal-field levels of 7F_2 : Γ_3^a at 940 cm^{-1} and Γ_3^b at 1061 cm^{-1} . The position of the Γ_1 level cannot be determined until now.

4.3. ${}^5D_2 \leftarrow {}^7F_{0,1,2}$

Transitions to the 5D_2 level are present in the spectrum between 20 000 and 21 600 cm^{-1} .

Two peaks in the α and σ spectrum belong to ${}^5D_2 \leftarrow {}^7F_0$: one at 21 461 cm^{-1} ($\Gamma_3^a \leftarrow \Gamma_1$) and another at 21 531 cm^{-1} ($\Gamma_3^b \leftarrow \Gamma_1$). No peaks are seen in π polarization. The transition has an ED character.

The transitions from 7F_1 to 5D_2 give four peaks in the σ spectrum: 21 000 cm^{-1} ($\Gamma_3^a \leftarrow \Gamma_2$), 21 071 cm^{-1} ($\Gamma_3^b \leftarrow \Gamma_2$), 21 130 cm^{-1} ($\Gamma_3^a \leftarrow \Gamma_3$) and 21 201 cm^{-1} ($\Gamma_3^b \leftarrow \Gamma_3$). The π spectrum shows no clear signals.

The ${}^5D_2 \leftarrow {}^7F_2$ transition gives a single peak in π polarization at 20 521 cm^{-1} ($\Gamma_3^a \leftarrow \Gamma_3^a$). Three signals are detected in the σ spectrum at 20 381 cm^{-1} ($\Gamma_1 \leftarrow \Gamma_3^b$) and at 20 501 cm^{-1} ($\Gamma_1 \leftarrow \Gamma_3^a$). The third peak at 20 241 cm^{-1} cannot be assigned unambiguously.

4.4. ${}^5D_3 \leftarrow {}^7F_{0,1}$

Transitions to 5D_3 are detected between 23 900 and 24 550 cm^{-1} . In π polarization, two peaks can be assigned: one at 24 327 cm^{-1} ($\Gamma_2^a \leftarrow \Gamma_1$) and one at 24 420 cm^{-1} ($\Gamma_2^b \leftarrow \Gamma_1$). The σ spectrum shows one peak at 24 357 cm^{-1} ($\Gamma_3^a \leftarrow \Gamma_1$).

The three signals of the ${}^5D_3 \leftarrow {}^7F_1$ transition can be assigned as follows: 23 946 cm^{-1} ($\Gamma_1 \leftarrow \Gamma_2$), 24 000 cm^{-1} ($\Gamma_3^a \leftarrow \Gamma_3$) and 24 020 cm^{-1} ($\Gamma_3^b \leftarrow \Gamma_3$). The absorption peak at 24 022 cm^{-1} in both α and σ spectra is the $\Gamma_2^a \leftarrow \Gamma_3$ transition and the peak at 24 075 cm^{-1} is the $\Gamma_2^b \leftarrow \Gamma_3$ one.

4.5. ${}^5L_6 \leftarrow {}^7F_{0,1}$

Transitions to 5L_6 are seen between 24 500 and 25 500 cm^{-1} . Because of the large splitting of the 5L_6 manifold (nearly 500 cm^{-1}), transitions from 7F_0 to the lowest crystal-field levels of 5L_6 are found in the same region as the transitions from 7F_1 to the highest crystal-field levels of 5L_6 . This ambiguity can be removed by recording spectra at 77 K, so that the thermal population of the 7F_1 crystal-field levels is very low and only transitions starting from the ground state 7F_0 are detected.

For the ${}^5L_6 \leftarrow {}^7F_0$ transition, four peaks are predicted theoretically in α and σ polarization and two peaks in π polarization. The two peaks in π signals are indeed found: one at 24 903 cm^{-1} ($\Gamma_2^a \leftarrow \Gamma_1$) and one at 25 231 cm^{-1} ($\Gamma_2^b \leftarrow \Gamma_1$). In the α and σ spectrum only three of the four peaks are found: 24 974 cm^{-1} ($\Gamma_3^a \leftarrow \Gamma_1$), 25 052 cm^{-1} ($\Gamma_3^b \leftarrow \Gamma_1$) and 25 367 cm^{-1} ($\Gamma_3^d \leftarrow \Gamma_1$).

Four signals in the α and σ spectra at room temperature and three signals in the π spectrum are assigned to the ${}^5L_6 \leftarrow {}^7F_1$ transition.

4.6. ${}^5D_4 \leftarrow {}^7F_{0,1}$

Transitions from 7F_0 and 7F_1 to the 5D_4 level are observed between 27 000 and 27 700 cm^{-1} .

Three signals of the ${}^5D_4 \leftarrow {}^7F_0$ transition are seen in the σ and α spectrum: 27 574 cm^{-1} ($\Gamma_3^a \leftarrow \Gamma_1$), 27 582 cm^{-1} ($\Gamma_3^b \leftarrow \Gamma_1$) and 27 660 cm^{-1} ($\Gamma_3^c \leftarrow \Gamma_1$). The π spectrum shows two peaks, one at 27 591 cm^{-1} ($\Gamma_2 \leftarrow \Gamma_1$) and another at 27 655 cm^{-1} . It is not possible to assign this transition unequivocally.

Transitions from 7F_1 are only found in σ polarization.

4.7. *Transitions to the other multiplets*

The assignments of the signals in the other spectral regions are not possible without the aid of calculated crystal-field levels (see further). A huge number of levels are calculated, but only a limited number of them are effectively found. Moreover, the overlap between the crystal-field levels of different spin-orbit coupling terms is very strong. This is also found for Eu^{3+} in other systems, e.g. EuODA [13] and LiYF_4 [14]. Between 26 000 and 29 000 cm^{-1} transitions to the 5L_J ($J = 7, 8, 9, 10$) and the 5G_J ($J = 2, 3, 4, 5, 6$) are found, whereas the 5H_J ($J = 3, 4, 5, 6, 7$) levels are between 29 000 and 32 000 cm^{-1} . Between 32 000 and 33 000 cm^{-1} the Gd^{3+} multiplets ${}^6P_{7/2}$ and ${}^6P_{5/2}$ are detected. Over 33 000 cm^{-1} , the 4f-4f transitions are overwhelmed by a broad intense band (charge-transfer band?).

All the observed transitions in the polarized absorption of Eu^{3+} -doped $\text{GdAl}_3(\text{BO}_3)_4$ are summarized in table 2. The spectra at 298 K are given in figure 1. It is obvious that the absorption peaks are sharp even at this temperature. This is a characteristic feature of the ions in the middle of the lanthanide series (Eu^{3+} , Gd^{3+} and Tb^{3+}).

Table 2. Summary of the transitions in the polarized absorption spectra of Eu^{3+} in $\text{GdAl}_3(\text{BO}_3)_4$. Assignments are made in a D_3 symmetry.

Transition	Assignment	Energy (cm^{-1})	Polarization	Transition	Assignment	Energy (cm^{-1})	Polarization
${}^5D_0 \leftarrow {}^7F_2$	$\Gamma_1 \leftarrow \Gamma_3^b$	16184	$\alpha + \sigma$	${}^5L_7 \leftarrow {}^7F_0$	$\Gamma_3 \leftarrow \Gamma_1$	25956	$\alpha + \sigma$
	$\Gamma_1 \leftarrow \Gamma_3^a$	16305	$\alpha + \sigma$		${}^5G_2 \leftarrow {}^7F_0$	$\Gamma_2 \leftarrow \Gamma_1$	26025
${}^5D_0 \leftarrow {}^7F_1$	$\Gamma_1 \leftarrow \Gamma_2$	16787	σ	$\Gamma_3 \leftarrow \Gamma_1$		26051	$\alpha + \sigma$
	$\Gamma_1 \leftarrow \Gamma_3$	16915	$\alpha + \pi$	$\Gamma_3 \leftarrow \Gamma_1$		26093	$\alpha + \sigma$
${}^5D_1 \leftarrow {}^7F_2$	$\Gamma_3 \leftarrow \Gamma_3^b$	17938	$\alpha + \sigma$	$\Gamma_2 \leftarrow \Gamma_1$		26179	π
	$\Gamma_2 \leftarrow \Gamma_3^b$	17969	$\alpha + \sigma$	$\Gamma_3 \leftarrow \Gamma_1$		26227	$\alpha + \sigma$
	$\Gamma_3 \leftarrow \Gamma_3^a$	18058	$\alpha + \sigma$	$\Gamma_3 \leftarrow \Gamma_1$	26272	$\alpha + \sigma$	
	$\Gamma_2 \leftarrow \Gamma_3^a$	18090	$\alpha + \sigma$	$\Gamma_2 \leftarrow \Gamma_1$	26288	π	
${}^5D_1 \leftarrow {}^7F_1$	$\Gamma_3 \leftarrow \Gamma_2$	18539	$\alpha + \sigma$	${}^5G_{3,4,5,6} \leftarrow {}^7F_0$	$\Gamma_3 \leftarrow \Gamma_1$	26435	$\alpha + \sigma$
	$\Gamma_3 \leftarrow \Gamma_3$	18667	$\alpha + \sigma$		$\Gamma_2 \leftarrow \Gamma_1$	26433	π
	$\Gamma_2 \leftarrow \Gamma_3$	18700	$\alpha + \sigma$		$\Gamma_3 \leftarrow \Gamma_1$	26455	$\alpha + \sigma$
${}^5D_1 \leftarrow {}^7F_0$	$\Gamma_3 \leftarrow \Gamma_1$	19000	$\alpha + \pi$		$\Gamma_2 \leftarrow \Gamma_1$	26465	π
	$\Gamma_2 \leftarrow \Gamma_1$	19030	σ		$\Gamma_3 \leftarrow \Gamma_1$	26484	$\alpha + \sigma$
${}^5D_2 \leftarrow {}^7F_2$	$\Gamma_1 \leftarrow \Gamma_3^b$	20381	σ		$\Gamma_2 \leftarrow \Gamma_1$	26513	π
	$\Gamma_1 \leftarrow \Gamma_3^a$	20501	σ	$\Gamma_3 \leftarrow \Gamma_1$	26517	$\alpha + \sigma$	
	$\Gamma_3^a \leftarrow \Gamma_3^b$	20521	π	$\Gamma_3 \leftarrow \Gamma_1$	26544	$\alpha + \sigma$	
${}^5D_2 \leftarrow {}^7F_1$	$\Gamma_3^a \leftarrow \Gamma_2$	21000	$\alpha + \sigma$	$\Gamma_3 \leftarrow \Gamma_1$	26583	$\alpha + \sigma$	
	$\Gamma_3^b \leftarrow \Gamma_2$	21071	$\alpha + \sigma$	$\Gamma_2 \leftarrow \Gamma_1$	26622	π	
	$\Gamma_3^a \leftarrow \Gamma_3$	21130	$\alpha + \sigma$	$\Gamma_3 \leftarrow \Gamma_1$	26650	$\alpha + \sigma$	
	$\Gamma_3^b \leftarrow \Gamma_3$	21201	$\alpha + \sigma$	$\Gamma_3 \leftarrow \Gamma_1$	26717	$\alpha + \sigma$	
${}^5D_2 \leftarrow {}^7F_0$	$\Gamma_3^a \leftarrow \Gamma_1$	21461	$\alpha + \sigma$	${}^5D_4 \leftarrow {}^7F_1$	$\Gamma_3^a \leftarrow \Gamma_2$	27115	σ
	$\Gamma_3^b \leftarrow \Gamma_1$	21531	$\alpha + \sigma$		$\Gamma_3^b \leftarrow \Gamma_2$	27198	σ
${}^5D_3 \leftarrow {}^7F_1$	$\Gamma_1 \leftarrow \Gamma_2$	23946	π		$\Gamma_3^a \leftarrow \Gamma_3$	27247	σ
	$\Gamma_3^a \leftarrow \Gamma_3$	24000	π		$\Gamma_3^b \leftarrow \Gamma_3$	27320	σ
	$\Gamma_3^b \leftarrow \Gamma_3$	24020	π	${}^5D_4 \leftarrow {}^7F_0$	$\Gamma_3^a \leftarrow \Gamma_1$	27574	$\alpha + \sigma$
	$\Gamma_2 \leftarrow \Gamma_3$	24022	$\alpha + \sigma$		$\Gamma_3^b \leftarrow \Gamma_1$	27582	$\alpha + \sigma$
	$\Gamma_2^b \leftarrow \Gamma_3$	24075	$\alpha + \sigma$		$\Gamma_2 \leftarrow \Gamma_1$	27591	π
${}^5D_3 \leftarrow {}^7F_0$	$\Gamma_2^a \leftarrow \Gamma_1$	24327	π	$\Gamma_1^b \leftarrow \Gamma_1$	27655	π	
	$\Gamma_3 \leftarrow \Gamma_1$	24357	σ	$\Gamma_3^a \leftarrow \Gamma_1$	27660	$\alpha + \sigma$	
	$\Gamma_3^b \leftarrow \Gamma_1$	24420	π	${}^5H_J \leftarrow {}^7F_0$	$\Gamma_3 \leftarrow \Gamma_1$	30668	$\alpha + \sigma$
${}^5L_6 \leftarrow {}^7F_1$	$\Gamma_1 \leftarrow \Gamma_2$	24514	π		$\Gamma_2 \leftarrow \Gamma_1$	30712	π
	$\Gamma_2^a \leftarrow \Gamma_2$	24523	$\alpha + \sigma$		$\Gamma_2 \leftarrow \Gamma_1$	30824	π
	$\Gamma_2^b \leftarrow \Gamma_2$	24570	$\alpha + \sigma$		$\Gamma_3 \leftarrow \Gamma_1$	30865	$\alpha + \sigma$
	$\Gamma_1 \leftarrow \Gamma_3$	24634	$\alpha + \sigma$		$\Gamma_2 \leftarrow \Gamma_1$	30973	π
	$\Gamma_1 \leftarrow \Gamma_3$	24723	$\alpha + \sigma$		$\Gamma_2 \leftarrow \Gamma_1$	31012	π
${}^5L_6 \leftarrow {}^7F_0$	$\Gamma_3^a \leftarrow \Gamma_3$	24835	π		$\Gamma_3 \leftarrow \Gamma_1$	31103	$\alpha + \sigma$
	$\Gamma_3^b \leftarrow \Gamma_3$	24903	π		$\Gamma_3 \leftarrow \Gamma_1$	31155	$\alpha + \sigma$
${}^5L_6 \leftarrow {}^7F_1$	$\Gamma_2^a \leftarrow \Gamma_1$	24915	$\alpha + \sigma$		$\Gamma_2 \leftarrow \Gamma_1$	31156	π
${}^5L_6 \leftarrow {}^7F_0$	$\Gamma_2^b \leftarrow \Gamma_1$	24974	$\alpha + \sigma$		$\Gamma_2 \leftarrow \Gamma_1$	31190	π
${}^5L_6 \leftarrow {}^7F_1$	$\Gamma_3^a \leftarrow \Gamma_2$	25031	π		$\Gamma_3 \leftarrow \Gamma_1$	31288	$\alpha + \sigma$
${}^5L_6 \leftarrow {}^7F_0$	$\Gamma_3^b \leftarrow \Gamma_1$	25052	$\alpha + \sigma$		$\Gamma_3 \leftarrow \Gamma_1$	31327	$\alpha + \sigma$
	$\Gamma_2^a \leftarrow \Gamma_1$	25231	π	$\Gamma_2 \leftarrow \Gamma_1$	31332	π	
	$\Gamma_2^b \leftarrow \Gamma_1$	25367	$\alpha + \sigma$	$\Gamma_2 \leftarrow \Gamma_1$	31400	π	
				$\Gamma_3 \leftarrow \Gamma_1$	31406	$\alpha + \sigma$	
				$\Gamma_3 \leftarrow \Gamma_1$	31537	$\alpha + \sigma$	
				$\Gamma_2 \leftarrow \Gamma_1$	31620	π	

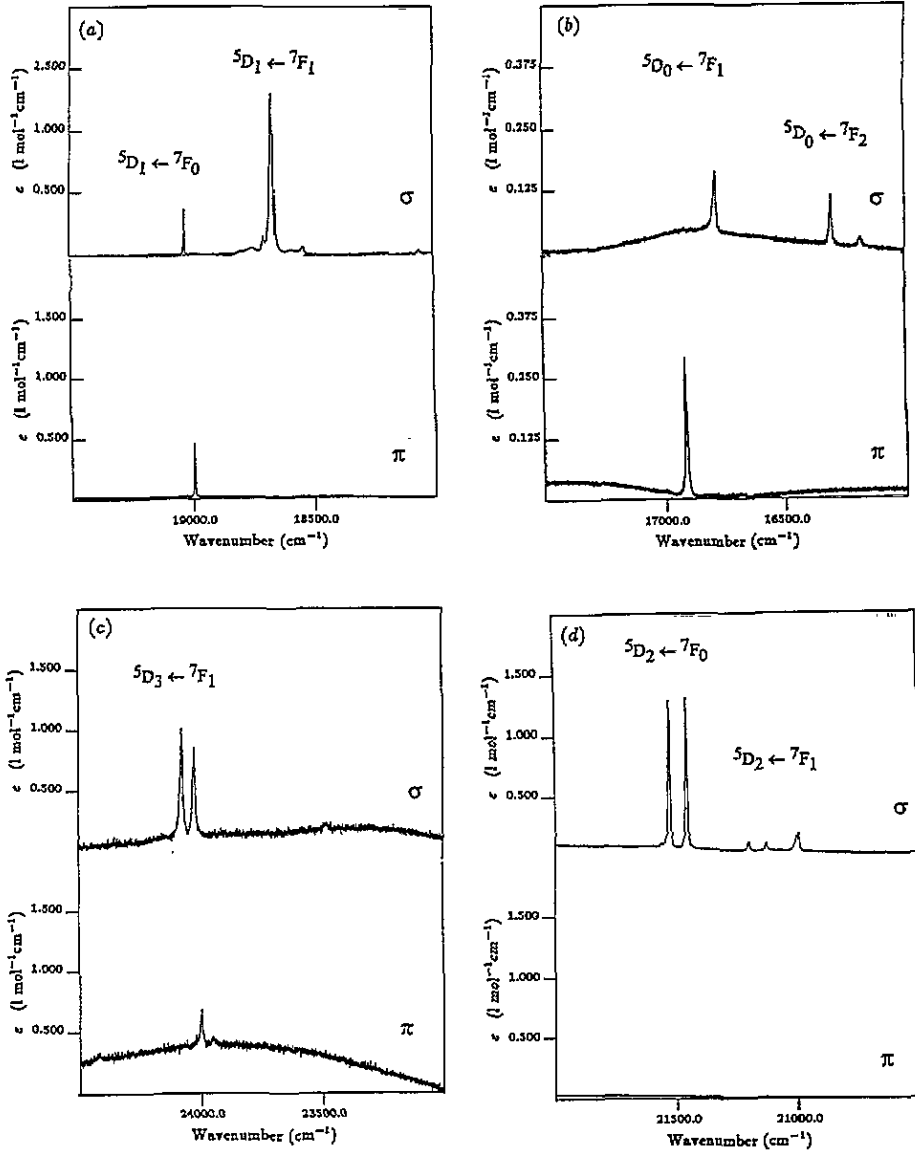


Figure 1. The polarized orthoaxial absorption spectra of Eu^{3+} -doped $\text{GdAl}_3(\text{BO}_3)_4$ at 298 K: (a) ${}^5\text{D}_1 \leftarrow {}^7\text{F}_0$, ${}^5\text{D}_1 \leftarrow {}^7\text{F}_1$; (b) ${}^5\text{D}_0 \leftarrow {}^7\text{F}_1$, ${}^5\text{D}_0 \leftarrow {}^7\text{F}_2$; (c) ${}^5\text{D}_3 \leftarrow {}^7\text{F}_1$; (d) ${}^5\text{D}_2 \leftarrow {}^7\text{F}_0$, ${}^5\text{D}_2 \leftarrow {}^7\text{F}_1$; (e) ${}^5\text{G}_{3,4,5,6} \leftarrow {}^7\text{F}_0$, ${}^5\text{L}_7 \leftarrow {}^7\text{F}_0$, ${}^5\text{G}_2 \leftarrow {}^7\text{F}_0$; (f) ${}^5\text{L}_6 \leftarrow {}^7\text{F}_0$, ${}^5\text{L}_6 \leftarrow {}^7\text{F}_1$; (g) ${}^5\text{H}_{3,4,5,6,7} \leftarrow {}^7\text{F}_0$; (h) ${}^5\text{D}_4 \leftarrow {}^7\text{F}_0$, ${}^5\text{D}_4 \leftarrow {}^7\text{F}_1$.

5. Energy level scheme and parametrization

The experimental energy sequence is simulated using a parametric Hamiltonian operation adapted to the f^6 configuration of the Eu^{3+} ion, including intermediate coupling and J mixing. The total Hamiltonian can be written as a free-ion part and a crystal-field part:

$$H = H_{\text{free ion}} + H_{\text{crystal field}}. \quad (1)$$

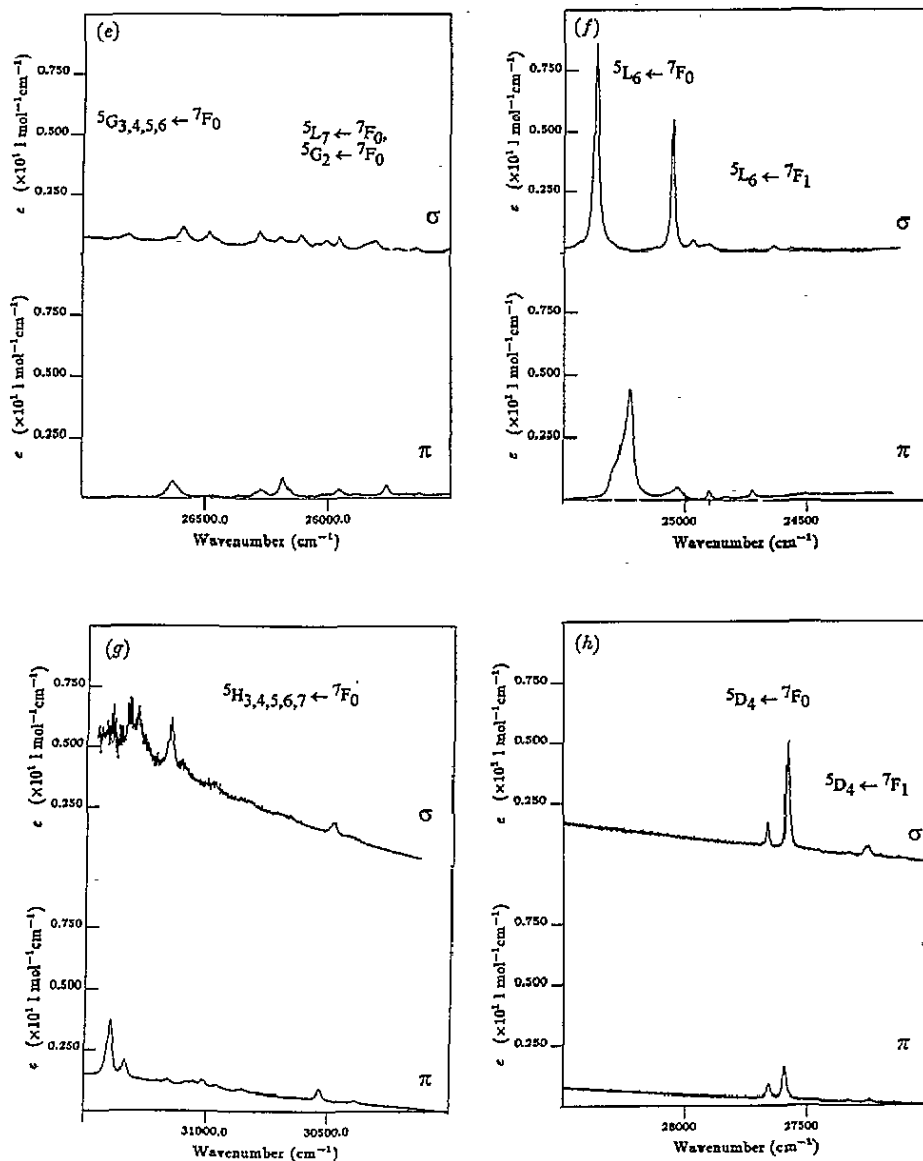


Figure 1. (Continued)

The free-ion Hamiltonian is characterized by a set of three repulsion parameters (F^2 , F^4 , F^6), by the spin-orbit coupling constant ζ_{4f} , the Trees configuration interaction parameters (α , β , γ), the three-body configuration-interaction parameters (T^2 , T^3 , T^4 , T^6 , T^7 , T^8) and parameters that describe magnetic interactions (M^0 , M^2 , M^4 , P^2 , P^4 , P^6). A further parameter E_{ave} takes into account the kinetic energy of the electrons and their interactions with the nucleus. It shifts only the barycentre of the whole $4f$ configuration. So one can write [15]

$$H_{\text{free ion}} = \sum_k F^k f_k + \zeta_{4f} A_{so} + \alpha L(L+1) + \beta G(G_2) + \gamma G(R_7) \\ + \sum_i T^i t_i + \sum_k P^k p_k + \sum_l M^l m_l \quad (2)$$

$i = 2, 3, 4, 6, 7, 8$; $k = 2, 4, 6$; $l = 0, 2, 4$.

Here f_k and A_{so} represent the angular part of the electrostatic and spin-orbit interaction respectively; L is the total orbital angular momentum; $G(G_2)$ and $G(R_7)$ are the Casimir operators for the groups G_2 and R_7 ; the t_i are the three-particle operators; and p_k and m_l represent the operators for the magnetic corrections.

The crystal-field Hamiltonian is given in the formalism of Wybourne [16] by

$$H_{\text{crystal field}} = \sum_{i=0}^N \sum_{k=0}^{\infty} \sum_{q=-k}^k B_q^k C_q^k(i). \quad (3)$$

Here $C_q^k(i)$ is a spherical operator of rank k , with components q . The B_q^k are the crystal-field parameters. N is the number of electrons, and i represents the i th electron. For f electrons $k \leq 6$. The non-zero conditions for the B_q^k are associated with the point-group site symmetry of the lanthanide ion. Only parameters with even k values are responsible for the crystal-field splitting. For a D_3 symmetry, the even part of $H_{\text{crystal field}}$ is expanded as

$$H_{D_3}^{\text{even}} = B_0^2 C_0^2 + B_0^4 C_0^4 + B_3^4 (C_{-3}^4 - C_3^4) + B_0^6 C_0^6 + B_3^6 (C_{-3}^6 - C_3^6) + B_6^6 (C_{-6}^6 + C_6^6). \quad (4)$$

The free-ion and crystal-field parameters are determined by optimizing a preliminary set of parameters in a least-squares fit to a set of experimentally determined crystal-field energy levels. The starting free-ion parameters are taken from $\text{LaF}_3:\text{Eu}^{3+}$ [17], and the starting crystal-field parameters are those from $\text{YAl}_3(\text{BO}_3)_4:\text{Eu}^{3+}$ [7]. J mixing is taken into account. First the crystal-field levels of 7F_0 , 7F_1 , 7F_2 , 5D_0 , 5D_1 , 5D_2 and 5L_6 are used in the fit. The parameters that are allowed to vary are F^2 , F^4 , F^6 , α , ζ_{4f} , B_q^4 and B_q^6 . The repulsion parameters were in the first instance, however, held in the hydrogen-like ratio [13].

$$F^4 = 0.668F^2 \quad F^6 = 0.494F^2.$$

The B_0^2 parameter was fixed at the value calculated from the splitting of the 7F_1 and 5D_1 levels. Indeed, the splitting of these levels is in the Russell-Saunders coupling scheme (7F_1 and 5D_1 have more than 95% Russell-Saunders character) only determined by that parameter. Using the energy matrix in a Russell-Saunders basis, one finds that

$$\Delta E({}^7F_1) = 0.30B_0^2 \text{ cm}^{-1} \quad \Delta E({}^5D_1) = 0.09B_0^2 \text{ cm}^{-1}.$$

Progressively more levels are introduced in the fitting procedure and more parameters were allowed to vary freely. The P^k parameters are always constrained to the ratio [17]

$$P^4 = 0.75P^2 \quad P^6 = 0.50P^2.$$

P^2 is allowed to freely vary.

The root-mean-square deviation (RMS) σ for the final fit was 12.8 cm^{-1} . The optimized parameter set is given in table 3. The experimental and calculated energy levels for Eu^{3+} in $\text{GdAl}_3(\text{BO}_3)_4$ can be found in table 4. A total of 68 crystal-field levels are located and assigned.

Table 3. Free-ion and crystal-field parameters (in a D_3 symmetry) for Eu^{3+} in $\text{GdAl}_3(\text{BO}_3)_4$. The RMS for the last fit is 12.8 cm^{-1} . The values in parentheses are the errors on the parameters.

Parameter	Value (cm^{-1})	Parameter	Value (cm^{-1})
E_{ave}	63 443	ζ_{4f}	1328.9(1.2)
F^2	81 787(60)	M^0	2.20
F^4	59 534(71)	M^2	1.23
F^6	43 017(40)	M^4	0.84
α	21.912	P^2	319
β	-656	P^4	0.75 P^2
γ	1377	P^6	0.50 P^2
T^2	415(12)	B_0^2	530(40)
T^3	104(180)	B_0^4	-1164(50)
T^4	130(32)	B_3^4	-945(70)
T^6	-49(185)	B_0^6	688(29)
T^7	289(66)	B_3^6	110(51)
T^8	458(30)	B_6^6	-518(49)

6. Sensitivity analysis

The optimized set of free-ion and crystal-field parameters (table 3) are considered to give the best agreement between experimental and calculated energy levels. One can ask if this set is a unique one or if there are more sets possible. Not all parameters can be determined very accurately, because some can be varied over a rather large interval without causing great changes in the position of the calculated crystal-field levels. Other parameters only have an influence on a limited number of levels. The influence of a +10% change of the optimized free-ion parameters on the position of the calculated free-ion levels is given in table 5. The free-ion levels were calculated by setting all crystal-field parameters equal to zero. Only one parameter is changed during a single calculation. From table 5 it can be deduced which levels are sensitive for which parameters. A small change of such a sensitive parameter will cause a great change of the calculated energy levels. Parameters that show a great sensitivity can be determined accurately. One can draw the following conclusions:

(i) The electron repulsion parameters F^k have little influence on the 7F multiplet, but the 5D and 5L multiplets are greatly influenced. A +10% change of F^2 causes a shift of 7F and 5D , which is opposite to the shift caused by F^4 and F^6 . The absolute value of the shift increases with increasing k value.

(ii) The spin-orbit coupling parameter ζ_{4f} has a small influence on the 5D_1 and 5L_6 , but most other levels are very sensitive to it. This parameter can be determined very accurately.

(iii) The configuration-interaction parameters α , β and γ have only little influence on the levels, compared to the electron repulsion on spin-orbit coupling parameters. The 5L_6 level is, however, rather sensitive to the α parameter, because α is important for levels with high L value (e.g. 5L_J , 5G_J , 5H_J). 5L_6 and 5D_J are sensitive to β and γ , but 5F_J is not.

(iv) The T^k parameters have a negligible effect on the lower energy levels ($< 25\,000 \text{ cm}^{-1}$). T^2 and T^4 have a distinct influence on 3P_J , and T^6 and T^7 on 5H_J .

(v) A change of M^k and P^k has no marked influence on the energy level calculation. They cannot be excluded from the calculation, because they are able to correct for minor deviations.

It is evident that parameters that have only an influence on levels that are not observed in the experiment cannot be determined. They must be held constant during the fitting

Table 4. Experimental and calculated energy levels for Eu^{3+} in $\text{GdAl}_3(\text{BO}_3)_4$. The levels are assigned in a D_3 symmetry.

$2S+1L_J$	Symmetry label	E_{exp} (cm $^{-1}$)	E_{calc} (cm $^{-1}$)	$2S+1L_J$	Symmetry label	E_{exp} (cm $^{-1}$)	E_{calc} (cm $^{-1}$)
7F_0	Γ_1	0	0	$^5G_{3,4,5,6}$	Γ_1	—	26 525
7F_1	Γ_3	331	334		Γ_1	—	26 557
	Γ_2	461	461	Γ_3	26 544	26 544	
7F_2	Γ_3^a	940	932	Γ_3	26 583	26 566	
		1061	1060	Γ_2	26 622	26 609	
	Γ_1	—	1195	Γ_3	26 650	26 634	
		—	—	Γ_1	—	26 635	
7F_3 – 7F_6	not observed in absorption	—	—	Γ_2	—	26 645	
				Γ_1	—	26 695	
				Γ_3	26 717	26 715	
				Γ_2	—	26 764	
				Γ_1	—	26 772	
5D_0	Γ_1	17 245	17 247	Γ_1	—	26 779	
5D_1	Γ_3	19 000	18 991	Γ_2	—	26 797	
	Γ_2	19 030	19 030	Γ_3	26 805	26 800	
5D_2	Γ_1	21 442	21 458	5D_4	Γ_3	26 834	26 842
		21 461	21 461		Γ_3^a	27 574	27 580
		21 531	21 513		Γ_1^a	—	27 582
5D_3	Γ_3^b	24 327	24 342	Γ_3^b	27 582	27 598	
		24 351	24 356	Γ_2^b	27 591	27 596	
		24 357	24 358	Γ_1^b	27 655	27 610	
		24 406	24 388	Γ_3^c	27 660	27 672	
		24 420	24 396	$^5H_{3,4,5,6,7}$	Γ_3	30 668	30 680
5L_6	—	—	—		Γ_2	30 712	30 699
					Γ_3	—	30 760
					Γ_1	—	30 781
					Γ_3	—	30 803
				Γ_2	30 824	30 797	
				Γ_1	—	30 846	
				Γ_3	30 865	30 860	
				Γ_2	30 973	30 972	
				Γ_2	31 012	31 011	
				Γ_1	—	31 028	
$^5L_7, ^5G_2$	—	—	—	Γ_3	—	31 065	
				Γ_3	—	31 077	
				Γ_1	—	31 069	
				Γ_3	31 103	31 115	
				Γ_2	25 956	25 943	
				Γ_3	26 025	26 041	
				Γ_3	26 051	26 048	
				Γ_1	—	26 068	
				Γ_3	26 093	26 087	
				Γ_2	26 179	26 183	
				Γ_3	26 227	26 225	
				Γ_3	26 272	26 260	
				Γ_1	—	26 277	
				Γ_2	26 288	26 298	
				Γ_3	—	26 309	
				Γ_3	26 332	26 357	
				Γ_1	—	26 383	
$^5G_{3,4,5,6}$	—	—	—	Γ_3	—	31 077	
				Γ_1	—	31 069	
				Γ_3	31 155	31 159	
				Γ_2	31 156	31 138	
				Γ_2	31 190	31 190	
				Γ_3	—	31 182	
				Γ_3	—	31 249	
				Γ_3	31 288	31 295	
				Γ_1	—	31 299	
				Γ_3	31 327	31 337	
Γ_1	—	31 329					
Γ_2	31 332	31 333					
Γ_3	—	31 357					
Γ_3	31 406	31 409					
Γ_1	—	31 412					
Γ_2	31 400	31 413					
Γ_3	31 537	31 525					
Γ_1	—	31 535					
Γ_1	—	31 542					
Γ_3	—	31 550					
Γ_3	—	31 569					
Γ_2	31 620	31 586					
Γ_3	—	—					

Table 5. Influence of a +10% change of the free-ion parameters on the calculated free-ion levels. The crystal-field parameters were set equal to zero. All values are expressed in cm^{-1} . Only one parameter was changed at a time.

ΔE	7F_1	7F_2	7F_3	7F_4	7F_5	7F_6	5D_0	5D_1	5D_2	5D_3	5D_4	5L_6
F^2	5.11	12.62	19.72	24.64	26.29	23.42	-267.21	-282.97	-284.64	-266.92	-208.48	409.84
F^4	-10.68	-23.10	-31.46	-33.95	-31.24	-24.64	994.95	1014.09	1030.16	1026.24	993.02	870.92
F^6	-18.92	-42.44	-61.12	-70.15	-66.51	-47.42	1277.46	1181.57	1139.89	1140.07	1181.74	975.29
ζ_{6f}	71.22	174.49	285.09	391.26	487.47	569.89	-325.18	-38.43	245.47	502.57	749.45	78.52
α	0.15	0.35	0.55	0.74	0.95	1.22	-10.00	-9.52	-9.22	-8.93	-8.16	97.84
β	0.75	1.63	2.25	2.47	2.25	1.63	-55.49	-53.54	-52.86	-53.19	-54.17	-69.54
γ	-1.72	-3.71	-5.09	-5.53	-4.97	-3.42	135.10	131.28	130.33	131.53	133.48	171.54
T^2	0.79	1.79	2.61	3.03	2.89	2.05	-46.39	-45.07	-44.54	-44.94	-45.86	-31.06
T^3	0.00	-0.01	-0.01	-0.01	-0.01	0.00	-0.03	0.03	0.08	0.09	0.08	-0.14
T^4	0.04	0.10	0.15	0.20	0.23	0.25	-2.67	-2.09	-1.70	-1.42	-1.25	-0.39
T^6	-0.02	-0.04	-0.07	-0.08	-0.09	-0.06	-0.25	-0.50	-0.79	-0.50	-0.12	-1.96
T^7	0.04	0.08	0.11	0.13	0.12	0.09	-2.26	-2.49	-2.62	-3.06	-3.11	-0.63
T^8	-0.23	-0.55	-0.84	-1.03	-1.05	-0.84	14.37	16.58	17.91	18.54	17.85	4.58
M^0	-1.24	-3.81	-7.76	-13.10	-19.88	-28.14	-8.49	-9.19	-11.56	-15.58	-20.88	7.84
M^2	-0.05	-0.07	-0.05	0.03	0.12	0.13	1.40	0.85	0.30	-0.27	0.71	-0.31
M^4	-0.02	-0.04	-0.04	-0.04	-0.04	-0.10	1.12	0.65	0.18	-0.29	-0.73	0.03
P^2	0.37	1.13	2.32	4.02	6.30	9.29	1.07	1.97	3.30	5.14	7.27	0.95
P^4	0.11	0.35	0.74	1.28	1.97	2.78	0.71	0.87	1.14	1.56	2.11	-0.92
P^6	0.01	0.04	0.09	0.17	0.28	0.45	0.76	0.47	0.26	0.10	-0.05	-0.20

procedure and are taken from parameter sets for the same ion in other matrices or can be found by extrapolation from parameter sets for other ions.

The sensitivity analysis can be extended to the crystal-field parameters. J mixing is applied in the calculations. Because of this J mixing, the free-ion levels (all crystal-field parameters set equal to zero) and the barycentres of the crystal-field levels do not coincide (table 6). J mixing has a great influence on the composition of the wavefunctions, and thus on the intensity of the transitions (see further). This was also noticed by Porcher and Caro [18] for Eu^{3+} in KY_3F_{10} . Because of the great number of crystal-field levels, only the influence of the parameters on ${}^5\text{D}_J$ and ${}^5\text{L}_6$ is considered.

Table 6. The calculated free-ion levels (all crystal-field parameters are set equal to zero) are not the same as the calculated barycentres of the crystal-field levels, when J mixing is taken into account. This is illustrated for the ${}^7\text{F}_J$, ${}^5\text{D}_J$ and ${}^5\text{L}_6$ levels of Eu^{3+} . All values are expressed in cm^{-1} .

$2S+1L_J$	Free-ion level	Barycentre	ΔE
${}^7\text{F}_1$	379	371	8
${}^7\text{F}_2$	1043	1014	29
${}^7\text{F}_3$	1898	1850	48
${}^7\text{F}_4$	2875	2861	14
${}^7\text{F}_5$	3924	3962	-38
${}^7\text{F}_6$	5012	5045	-33
${}^5\text{D}_0$	17260	17229	31
${}^5\text{D}_1$	19011	18984	27
${}^5\text{D}_2$	21486	21441	45
${}^5\text{D}_3$	24376	24265	111
${}^5\text{D}_4$	27620	27557	63
${}^5\text{L}_6$	25140	25052	88

The splitting of the ${}^5\text{D}_1$ is very sensitive to the B_0^2 parameter. The sign determines the relative position of the Γ_2 and Γ_3 levels. The magnitude of the parameter is proportional to the magnitude of the crystal-field splitting (as mentioned above). The B_0^2 parameter determines further the relative position of Γ_1 and Γ_2 in ${}^5\text{D}_3$ and of the Γ_3 levels in ${}^5\text{L}_6$. The B_0^4 parameter is important for the splitting of the ${}^5\text{D}_2$ level. The position of Γ_1 relative to the two Γ_3 levels is only determined by this parameter. The B_3^4 has a distinct influence on the magnitude of the crystal-field splitting of ${}^5\text{D}_2$. The B_0^6 parameter can be determined accurately using the ${}^5\text{D}_4$ crystal-field levels. The parameter has only a global influence on the ${}^5\text{D}_1$, ${}^5\text{D}_2$ and ${}^5\text{D}_3$ levels. This is also to a lesser extent true for the B_3^6 and B_6^6 parameters. The crystal-field splitting of levels with low J values is to a minor degree influenced by B_q^k parameters with high k values. All the B_q^k parameters are responsible for the splitting of the ${}^5\text{L}_6$ term.

7. Intensity parameters

The intensity of the $4f-4f$ transitions is measured in terms of the dipole strength, which can be deduced from the spectra by integrating the absorption peaks:

$$D_{\text{exp}} = \frac{1}{3 \times 108.9} \frac{g_i}{X_i(T)} \int \frac{\varepsilon(\bar{\nu})}{\bar{\nu}} d\bar{\nu} \quad (5)$$

where g_i is the degeneracy of the crystal-field ground level i , $X_i(T)$ is the fractional population of the crystal-field ground level i at temperature T , $\varepsilon(\bar{\nu})$ is the molar absorptivity at wavenumber $\bar{\nu}$ and D_{exp} is the experimental dipole strength (Debye²). D_{exp} contains both ED and MD contributions:

$$D_{\text{exp}} = \chi_{\text{MD}} D_{\text{exp}}^{\text{MD}} + \chi_{\text{ED}} D_{\text{exp}}^{\text{ED}} \quad (6)$$

χ_{MD} and χ_{ED} are corrections for medium effects:

$$\chi_{\text{MD}} = n \quad (7)$$

$$\chi_{\text{ED}} = (n^2 + 2)^2 / 9n \quad (8)$$

where n is the refractive index. For uniaxial crystals with a small birefringence, the two main refractive indices can be replaced by a mean value. For $\text{GdAl}_3(\text{BO}_3)_3$, $\chi_{\text{MD}} \simeq 1.75$ and $\chi_{\text{ED}} \simeq 1.63$. The dipole strengths are determined for the orthoaxial σ and π spectra.

The dipole strength D is also defined as the absolute square of the matrix element between the dipole operator and the crystal-field wavefunctions $\Psi_{\tau SLJM}$ and $\Psi_{\tau' SL'J'M'}$ of both the ground level and the excited level:

$$D = |\langle \Psi_{\tau SLJM} | O | \Psi_{\tau' SL'J'M'} \rangle|^2 \quad (9)$$

where O represents the MD operator μ_ρ or the ED operator m_ρ .

The total calculated dipole strength is given by

$$D_{\text{calc}} = \chi_{\text{MD}} D_{\text{calc}}^{\text{MD}} + \chi_{\text{ED}} D_{\text{calc}}^{\text{ED}} \quad (10)$$

The dipole strength of a MD transition can be calculated using the crystal-field wavefunctions which are generated by the energy parameters. Only a limited number of transitions have a predominant MD nature: ${}^5D_1 \leftarrow {}^7F_0$, ${}^5D_0 \leftarrow {}^7F_1$ and ${}^5D_2 \leftarrow {}^7F_1$.

However, for the calculation of the dipole strength of an ED transition, an additional set of parameters is required, the $B_{\lambda kq}$ intensity parameters:

$$\begin{aligned} & \langle \Psi_{\tau SLJM} | m_\rho | \Psi_{\tau' SL'J'M'} \rangle \\ &= -|e| \sum_{\lambda kq} B_{\lambda kq} (2\lambda + 1) \begin{pmatrix} 1 & \lambda & k \\ \rho & -q - \rho & q \end{pmatrix} (-1)^{q+\rho} \langle \Psi_{\tau SLJM} | U_{q+\rho}^\lambda | \Psi_{\tau' SL'J'M'} \rangle \\ &= -|e| \sum_{\lambda kq} a_{\lambda kq}^\rho B_{\lambda kq}. \end{aligned} \quad (11)$$

An explanation of the intensity model used can be found in Reid and Richardson [9–11]. These authors use, however, A_{tp}^λ parameters:

$$A_{tp}^\lambda = -B_{\lambda kq} (2\lambda + 1) / (2k + 1)^{1/2} \quad t = k \text{ and } p = q. \quad (12)$$

The $B_{\lambda kq}$ parameters are more suitable for crystal-field transitions than the traditional Judd–Ofelt parameters [19, 20], because the latter describe the intensity between whole spin–orbit coupling bands. The site symmetry of the lanthanide ion places certain restrictions on the permissible pairs of (k, q) for each λ . In a D_3 symmetry, one has to consider 21 different $B_{\lambda kq}$ intensity parameters. The allowed (λ, k, q) combinations are: $(2, 2, 0)$, $(2, 3, \pm 3)$,

(4,3,±3), (4,4,0), (4,4,±3), (4,5,±3), (6,5,±3), (6,6,0), (6,6,±3), (6,6,±6), (6,7,±3) and (6,7,±6). The number of parameters can be reduced to 12 using the relationship

$$B_{\lambda k q} = (-1)^{k+q+1} B_{\lambda k, -q}^* \quad (13)$$

The intensity parameters are determined by fitting the calculated $D_{\text{calc}}^{\text{ED}}$ against the experimental $D_{\text{exp}}^{\text{ED}}$ (which is determined from D_{exp} minus the calculated MD contributions). Not every $B_{\lambda k q}$ has an equal contribution to the intensity of an ED transition. In fact, the following selection rule is applicable:

$$|\Delta J| \leq \lambda \leq J + J' \quad (14)$$

The ${}^5D_2 \leftarrow {}^7F_0$ and ${}^5D_1 \leftarrow {}^7F_1$ transitions can be used for the determination of the B_{220} and B_{223} parameters. Transitions to the 5L_6 level are suitable for fitting the six B_{6xx} parameters. In theory only the B_{4xx} parameters are needed to calculate the intensity of transitions to the 5D_4 multiplet, but the B_{6xx} parameters have also a major contribution. Indeed, the wavefunctions of the 5D_4 crystal-field levels have a significant 5G_6 character. After the preliminary determination of the $B_{\lambda k q}$ parameters, the parameter set is refined by taking the other transitions into account.

The $B_{\lambda k q}$ parameters are given in table 7. There is an uncertainty about the sign of the intensity parameters. This has no effect on the calculated dipole strengths, but it may have an effect on other calculated parameters such as the rotatory strength and MCD parameters [21]. The calculated and experimental dipole strengths are summarized in table 8.

Table 7. $B_{\lambda k q}$ intensity parameters for Eu^{3+} in $\text{GdAl}_3(\text{BO}_3)_4$.

Parameter	Value (cm)
B_{220}	$61 \times 10^{-12}i$
B_{233}	$227 \times 10^{-12}i$
B_{433}	$104 \times 10^{-12}i$
B_{440}	$-216 \times 10^{-12}i$
B_{443}	$46 \times 10^{-12}i$
B_{453}	$-99 \times 10^{-12}i$
B_{653}	$30 \times 10^{-12}i$
B_{660}	$70 \times 10^{-12}i$
B_{663}	$2 \times 10^{-12}i$
B_{666}	$-5 \times 10^{-12}i$
B_{673}	$111 \times 10^{-12}i$
B_{676}	$-64 \times 10^{-12}i$

8. Conclusions

A detailed analysis of the absorption spectrum of Eu^{3+} -doped $\text{GdAl}_3(\text{BO}_3)_4$ is given. The energy levels are parametrized in terms of 20 free-ion and six crystal-field parameters. The spectral intensities are simulated using a set of $B_{\lambda k q}$ intensity parameters. Anisotropic ligand polarization is taken into account. This model gives a fairly good description of the f-f transition probabilities. The present work provides the first complete set of energy and intensity parameters for lanthanide ions surrounded by borate groups. The crystal-field

Table 8. Experimental and calculated dipole strengths of 4f–4f transitions for Eu^{3+} in $\text{CdAl}_3(\text{BO}_3)_4$. All values are expressed in 10^{-8} Debye². The ED intensities are calculated using the $B_{\lambda k q}$ parameters from table 7.

Transition	Symmetry label	Polarization	Mechanism	D_{exp}	D_{calc}
${}^5\text{D}_0 \leftarrow {}^7\text{F}_1$	$\Gamma_1 \leftarrow \Gamma_2$	σ	MD	557	536
	$\Gamma_1 \leftarrow \Gamma_3$	π	MD	471	540
${}^5\text{D}_1 \leftarrow {}^7\text{F}_0$	$\Gamma_3 \leftarrow \Gamma_1$	π	MD	105	123
	$\Gamma_2 \leftarrow \Gamma_1$	σ	MD	67	68
${}^5\text{D}_1 \leftarrow {}^7\text{F}_1$	$\Gamma_3 \leftarrow \Gamma_2$	σ	ED	189	272
	$\Gamma_2 \leftarrow \Gamma_2$	—	—	—	0.0
	$\Gamma_3 \leftarrow \Gamma_3$	σ	ED	3684	3583
	$\Gamma_3 \leftarrow \Gamma_3$	π	—	—	0.0
	$\Gamma_2 \leftarrow \Gamma_3$	σ	ED	181	257
${}^5\text{D}_2 \leftarrow {}^7\text{F}_0$	$\Gamma_1 \leftarrow \Gamma_1$	—	—	—	0.0
	$\Gamma_3^a \leftarrow \Gamma_1$	σ	ED	641	846
	$\Gamma_3^b \leftarrow \Gamma_1$	σ	ED	631	637
${}^5\text{D}_2 \leftarrow {}^7\text{F}_1$	$\Gamma_1 \leftarrow \Gamma_2$	—	—	—	3
	$\Gamma_3^a \leftarrow \Gamma_2$	σ	ED	504	285
	$\Gamma_3^b \leftarrow \Gamma_2$	—	ED	—	0.0
	$\Gamma_1 \leftarrow \Gamma_3$	—	—	—	0.0
	$\Gamma_3^a \leftarrow \Gamma_3$	σ	ED	62	5
	$\Gamma_3^b \leftarrow \Gamma_3$	σ	ED	75	103
${}^5\text{D}_3 \leftarrow {}^7\text{F}_0$	$\Gamma_2^a \leftarrow \Gamma_1$	π	ED	2	2
	$\Gamma_2^b \leftarrow \Gamma_1$	π	ED	5	0.4
	$\Gamma_1 \leftarrow \Gamma_1$	—	—	—	0.0
	$\Gamma_2^b \leftarrow \Gamma_1$	—	—	—	2
	$\Gamma_3^a \leftarrow \Gamma_1$	σ	ED	7	0.5
${}^5\text{L}_6 \leftarrow {}^7\text{F}_0$	$\Gamma_1^a \leftarrow \Gamma_1$	π	ED	—	0.0
	$\Gamma_2^a \leftarrow \Gamma_1$	π	ED	121	153
	$\Gamma_1^b \leftarrow \Gamma_1$	π	ED	—	0.0
	$\Gamma_1^c \leftarrow \Gamma_1$	π	ED	—	0.0
	$\Gamma_2^b \leftarrow \Gamma_1$	π	ED	3239	3601
	$\Gamma_3^a \leftarrow \Gamma_1$	σ	ED	258	466
	$\Gamma_3^b \leftarrow \Gamma_1$	σ	ED	2794	2818
	$\Gamma_3^c \leftarrow \Gamma_1$	σ	ED	—	197
	$\Gamma_3^d \leftarrow \Gamma_1$	σ	ED	5415	5183
	${}^5\text{D}_4 \leftarrow {}^7\text{F}_0$	$\Gamma_3^a \leftarrow \Gamma_1$	σ	ED	776
$\Gamma_3^b \leftarrow \Gamma_1$		σ	ED	680	681
$\Gamma_3^c \leftarrow \Gamma_1$		σ	ED	262	264
$\Gamma_2 \leftarrow \Gamma_1$		π	ED	466	463
$\Gamma_1^b \leftarrow \Gamma_1$		—	ED	238 ^a	0.0
$\Gamma_1^c \leftarrow \Gamma_1$		—	ED	—	0.0

^a A $\Gamma_1 \rightarrow \Gamma_1$ transition is forbidden by the selection rules in D_3 symmetry. Its presence is probably due to depolarization effects.

splittings are comparable to those found in the corresponding Y^{3+} compound [7]. Although the crystal-field felt by the Eu^{3+} in other matrices is expected to be different (compare with

EuODA [13] and $\text{LiYF}_4:\text{Eu}^{3+}$ [14]), there are also small changes in the free-ion parameters. This may be due, for example, to nephelauxetic effects [22]. The influence of a parameter change on the calculated energy levels is discussed. It is worth noting that some parameters have only a distinct influence on particular levels. A change of the repulsion parameters F^k has little influence on the ${}^7\text{F}$ levels, whereas ${}^5\text{D}$ and ${}^5\text{L}$ are greatly influenced. The spin-orbit coupling parameter ζ_{4f} has a strong influence on most levels, except ${}^5\text{D}_1$ and ${}^5\text{L}_6$. The ${}^5\text{L}_6$ is, on the other hand, rather sensitive to α . This is expected for a term with a high orbital angular momentum L . ${}^5\text{D}_J$ and ${}^5\text{L}_6$ are sensitive to β and γ , but ${}^7\text{F}_J$ is not. T^2 and T^4 have a distinct influence on ${}^3\text{P}_J$, T^6 and T^7 on ${}^5\text{H}_J$. Not all crystal-field parameters are responsible for the splitting of a particular J level. So the splitting of ${}^5\text{D}_1$ and ${}^7\text{F}_1$ is described in a good approximation only by the B_0^2 parameter.

Knowledge of this sensitivity is a valuable tool in the fitting procedure. It serves to decide which parameters may be varied freely and which may not. The calculated positions of those sensitive levels are a good check for the reliability of a parameter.

Acknowledgments

We thank H Crosswhite for developing the programs that are used in simulating the energy level scheme. KB (aspirant NFWO) thanks the NFWO (Belgium) for financial support. The Belgian Government is also gratefully acknowledged (Programmatie van het Wetenschapsbeleid).

References

- [1] Kuroda R, Mason S F and Rosini C 1981 *J. Chem. Soc. Faraday Trans. 2* **77** 2125
- [2] Lagerwey A A F and Blasse G 1975 *Chem. Phys. Lett.* **31** 27
- [3] Peacock R D 1975 *Chem. Phys. Lett.* **35** 420
- [4] Kellendonk F and Blasse G 1981 *J. Chem. Phys.* **75** 561
- [5] Görller-Walrand C and Vandeveldel P 1985 *Bull. Soc. Chim. Belg.* **94** 873
- [6] Görller-Walrand C and Vandeveldel P 1985 *Chem. Phys. Lett.* **122** 276
- [7] Görller-Walrand C, Vandeveldel P, Hendrickx I, Porcher P, Krupa J C and King G S D 1988 *Inorg. Chim. Acta* **143** 259
- [8] Mason S F, Peacock R D and Stewart B 1975 *Mol. Phys.* **30** 1829
- [9] Reid M F and Richardson F S 1983 *Chem. Phys. Lett.* **95** 501
- [10] Reid M F and Richardson F S 1983 *J. Chem. Phys.* **79** 5735
- [11] Reid M F and Richardson F S 1984 *J. Phys. Chem.* **88** 3579
- [12] Koster G F, Dimmock J O, Wheeler R G and Statz H 1963 *Properties of the Thirty-two Point Groups* (Cambridge, MA: MIT)
- [13] Berry M T, Schwieters C and Richardson F S 1988 *Chem. Phys.* **122** 105
- [14] Görller-Walrand C, Binnemans K and Fluyt L 1993 *J. Phys.: Condens. Matter* **5** 8359
- [15] Crosswhite H M and Crosswhite H 1984 *J. Opt. Soc. Am. B* **1** 246
- [16] Wybourne B G 1965 *Spectroscopic Properties of Rare Earths* (New York: Wiley)
- [17] Carnall W T, Goodman G L, Rajnak K and Rana R S 1988 A systematic analysis of the spectra of the lanthanides doped into single crystal LaF_3 *Technical Report* Argonne National Laboratory
- [18] Porcher P and Caro P 1980 *J. Lumin.* **21** 207
- [19] Judd B R 1962 *Phys. Rev.* **127** 750
- [20] Ofelt G S 1962 *J. Chem. Phys.* **37** 511
- [21] Görller-Walrand C, Verhoeven P, Fluyt L, D'Olieslager J and Binnemans K 1994 *J. Chem. Phys.* **100** 815
- [22] Caro P, Beaury O and Antic E 1976 *J. Physique* **37** 671